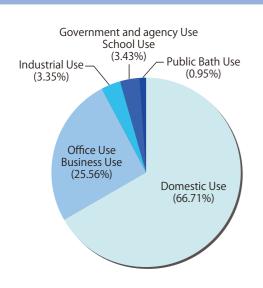
# Waterworks Date

### Waterworks Operations

**Commence of Operations** 

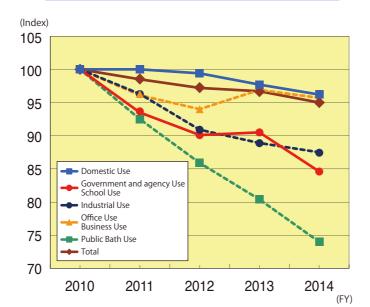
November 13, 1895 (the fourth modern water system in Japan after those of Yokohama, Hakodate, and Nagasaki)

- **Population Served** 2,690,214 (As of April 1, 2015)
- Number of Households Served 1,536,275 (As of March 31, 2015)
- Water Supply Coverage 100%
- Daily Water Supply Capacity 2,430,0000m<sup>3</sup>
  - **Settled Water Consumption Rates**

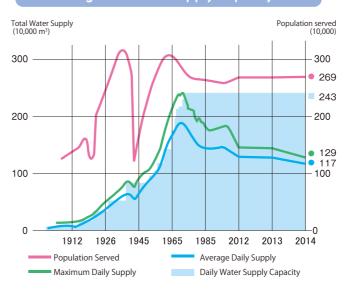


by Business Category for FY 2014

- **Annual Water Supply** 426,432,700m<sup>3</sup> (FY 2014))
- Maximum Daily Supply 1,286,700m<sup>3</sup> (FY 2014)) (The largest maximum daily supply recorded in 1970)
- **Average Daily Supply** 1.168.309m3 (FY 2014))
- Average Domestic Daily Water Consumption per Persón 243 l (FY 2014))
- Total length of aqueduct, water pipe and distributing pipe 5,230km (As of March 31, 2015)
- Annual Waterworks Budget 99,977,000,000 yen (Budget for FY 2015)
- **Settled Water Consumption** Index by Business Category



#### Population Served / Total Water Supply Changes in Water Supply Capacity



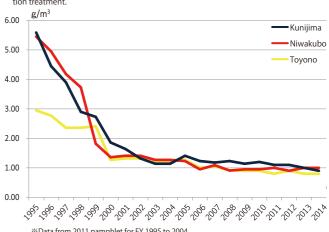
### Results of FY 2014 Osaka City Vater Quality Testing

#### Water Quality Standard Items Standard value 2 E coli hacteria Must not be detected \* 3. Cadmium and its compou 0.003mg/l or les less than 0.003 mg/l 4. Mercury and its compounds 0.0005mg/l or les less than 0.00005 mg/l \* 5. Seleium and its compounds Lead and its compounds 0.01mg/l or les \* 7. Arsenic and its compounds 0.01mg/l or les less than 0.0005 mg/l 8. Hexavalent chrome and its compounds 0.05mg/l or les less than 0.005 mg/l Nitrite nitrogen 0.04mg/l or les 10. Cyanide ion and cyanogen chloride 0.01mg/l or les 11. Nitrite nitrogen and nitrate nitrogen 12. Fluoride and its compounds 10mg/l or les 0.9 mg/l \* 13. Boron and its compounds 1.0ma/l or les 0.02 mg/l \* 15. 1,4-dioxane 0.05mg/l or les less than 0.0005 mg/l 0.04mg/lor less \* 17 Dichloromethane 0.02mg/lor less less than 0.001 mg/l \* 19. Trichloroethylen 0.01ma/l or les less than 0.003 mg/l 21. Chloric acid 0.6mg/l or les 0.026 mg/l 22. Chloroacetic acid 23. Chloroform 0.06mg/l or les 0.001 mg/l 24 Dichloroacetic acid less than 0.001 mg/ 25. Dibromochlo 0.1mg/l or les 0.005 mg/l 26. Bromic acid 27. Total trihalom 0.01mg/l or les 0.1mg/l or les 0.001 mg/ 28. Trichloroacetic acid 0.2mg/l or les less than 0.001 mg/l 29. Bromodichloromet 0.03mg/l or les 0.003 mg/l 30. Bromoform 0.09mg/l or les 0.08mg/l or les 0.002 mg/ 32, Zinc and its compounds 1.0ma/l or les less than 0.1 mg/l 33. Aluminum and its compou less than 0.01 mg/l 34. Iron and its compounds 0.3mg/l or les 35. Copper and its compounds 36. Sodium and its compounds 200mg/l or les 37. Manganese and its compounds 0.05mg/l or les ess than 0.001 mg/l \* 39. Calcium, magnesium etc. (hardness) \* 40. Evaporated residue 300mg/l or les 41 mg/l 500mg/l or les \* 41. Anionic surfactants 0.2mg/l or les less than 0.02 mg/l 0.00001mg/l or les less than 0.000001 mg/l less than 0.02 mg/l 42. Geosmin 43. 2-Methyl-isoborneol 0.00001mg/l or les less than 0.000001 mg/l \* 45. Phenols 0.005mg/l or les less than 0.0005 mg/l 3mg/l or les 5.8~8.6 47. pH value Must not be abnorma 49. Odor Must not be abnormal Chlorine odor

Asterisked items show values measured in the purification plant outflow. All other values were measured at 21 tap locations within the city.
 The value of total trihalomethanes is the total concentration value of chloroform, dibromochioromethane, bromodichloromethane, and bromoform.

#### Reduction of Chlorine Dosage Rate at the Purification Plants

The amount of chlorine used in the chlorination process has been greatly reduced by improvements in river water quality and the introduction of advanced water purifica

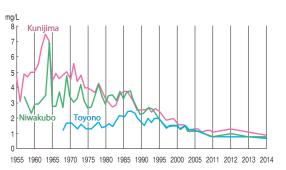


\*\*Data from 2011 pamphlet for FY 1995 to 2004
For FY 2005 to 2014, culculated based on data from Osaka City Council's settlement results.

#### BOD

#### [Biochemical Oxygen Demand (BOD)]

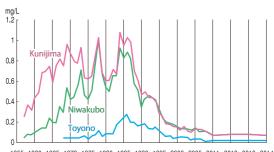
BOD is a biochemical procedure for determining the amount of oxygen required to decompose organic matter by microorganisms in a volume of water. A larger BOD figure signifies that a larger volume of organic substances is present, thus indicating greater



## Ammonia nitrogen

#### [Ammonia Nitrogen]

Ammonia nitrogen is a compound found in industrial effluents and raw wastewater. Its concentration in water is used as an indicator of water contamination and higher concentrations indicate higher water contamination values. High concentrations of ammonia nitrogen in the source water at purification plants require larger doses of chlorine for disinfection.



1955 1960 1965 1970 1975 1980 1985 1990 1995 2000 2005 2011 2012 2013 2014